

단 신

오메가-(메틸sulfinyl)아세토펜류를 이용한 2-아릴벤조푸란 유도체의 합성

金永和 · 崔洪大* · 徐弼子 · 孫炳華†

동의대학교 자연과학대학 화학과

†부경대학교 자연과학대학 화학과

(2001. 5. 31 접수)

Synthesis of 2-Arylbenzofuran Derivatives Using ω -(Methylsulfinyl)acetophenones

Young-Wha Kim, Hong-Dae Choi*, Pil-Ja Seo, and Byeng-Wha Son†

Department of Chemistry, Donggeui University, Pusan 614-714, Korea

†Department of Chemistry, Pukyong National University, Pusan 608-737, Korea

(Received May 31, 2001)

A series of benzofuran ring system bearing various substituents at the C-2 position is widely distributed in nature and has recently become of interest in biological properties. There are well known natural products having related benzofuran ring structures, particularly those isolated from *Machilus glaucescens*,¹ *Ophryosporus charua*,² *Ophryosporus lorentzii*,³ *Krameria ramosissima*,⁴ and *Zanthoxylum ailanthoidol*.⁵

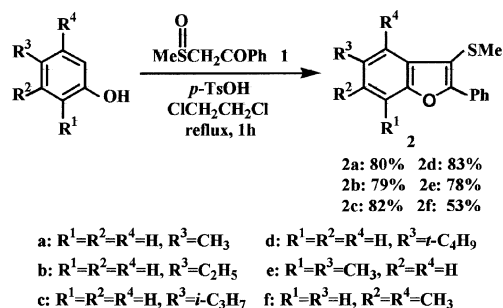
For the carbon-carbon bond formation using 1-acyl-1-thiocarbocations, we found that 2-methyl- and arylbenzofurans were easily prepared by the one-pot reaction of substituted phenols with 1-acyl-1-chlorosulfides in the presence of a Lewis acid.⁶ Also a facile one-pot procedure was developed, which offers 2-alkylbenzofurans from substituted phenols using α -acylsulfoxides in the presence of *p*-toluenesulfonic acid.⁷

In this paper, we report a new route for synthesizing of 2-arylbenzofuran derivatives from substituted phenols with α -acylsulfoxides (**1,3,4**) under Pummerer reaction conditions.

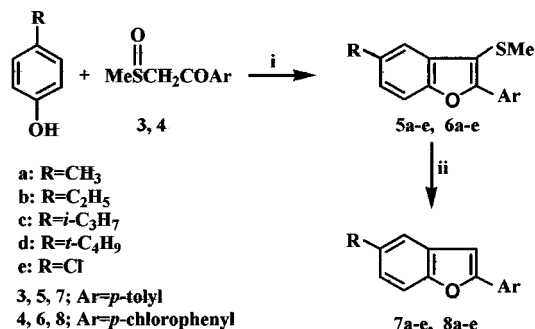
As preparation of the starting materials, ω -(methylsulfinyl)acetophenone (**1**) was obtained from the oxidation of 2-(methylthio)acetophenone with sodium meta-periodate in aqueous methanol in 80% yield. The reactions of ethyl *p*-toluate and ethyl *p*-chlorobenzoate with

methylsulfinyl carbanion⁸ afforded ω -methylsulfinyl-*p*-methylacetophenone (**3**) and ω -methylsulfinyl-*p*-chloroacetophenone (**4**) in 74% and 79% yields, respectively.

On the basis of our synthetic method⁷ for 2-methylbenzofurans using α -(methylsulfinyl)acetone under Pummerer reaction conditions, we first attempted the synthesis of 2-phenylbenzofurans **2** as illustrated in Scheme 1. Thus, treatment of equimolar amounts of substituted phenols and the sulfoxide **1** in 1,2-dichloroethane with three equivalents of anhydrous *p*-toluenesulfonic acid under reflux gave the compounds **2a-f** in moderate yields. The spectroscopic data (mp, IR, and ¹H NMR) were in good agreements with those reported by our pre-



Scheme 1



Scheme 2. Reagents and conditions: (i) *p*-TsOH, ClCH₂CH₂Cl, reflux, 1 h; (ii) Raney-Ni (W-2), EtOH, 60–65 °C, 1 h.

vious work^{6b} on the synthesis of 2-methylthio-2-phenylbenzofurans under Friedel-Crafts reaction conditions. The procedure for desulfurization of the adducts **2a–f** by heating with Raney nickel in ethanol was reported previously.^{6b}

Secondly, we applied the above method to syntheses of 3-methylthio-2-(*p*-tolyl)benzofurans (**5**) and 3-methylthio-2-(*p*-chlorophenyl)benzofurans (**6**), in which the sulfoxides **3** and **4** are employed as electrophiles in place of **1**. The Pummerer reactions of *para*-substituted phenols with **3** and **4** were carried out as shown in Scheme 2.

The treatment of equimolar amounts of *para*-substituted phenols and the sulfoxide **3** in the presence of three equivalents of *p*-toluenesulfonic acid afforded the compounds **5a–e** in satisfactory yields. Also the compounds **6a–e** were obtained from the reactions of *para*-substituted phenols and the sulfoxide **4** in the presence of *p*-toluenesulfonic acid.

The adducts (**5,6**) given by the above Pummerer reaction can easily be desulfurized into the corresponding 2-arylbenzofurans (**7,8**) by heating Raney nickel in ethanol. Thus, the adducts **5a–e** and **6a–e** were converted into 2-(*p*-tolyl)benzofurans **7a–e** and 2-(*p*-chlorophenyl)benzofurans **8a–e**, respectively, in high yields.

Of the many methods for the preparation of 2-arylbenzofuran ring, the route⁹ through the coupling reaction of an *o*-halophenol with a cuprous arylacetylide have been regarded as an efficient procedure. This method requires uncommon starting materials and lengthy reaction time.

In conclusion, we developed a new one-pot method for the construction of 2-arylbenzofurans (**2,5,6**) using

substituted phenols and α -acylsulfoxides (**1,3,4**) in the presence of anhydrous *p*-toluenesulfonic acid. This method is generally applicable to benzofuran moiety having various aryl groups at the C-2 position.

The Pummerer reactions for utilizing α -acylsulfoxides has been proved to be useful to synthesize the naturally occurring products possessing 2-arylbenzofuran skeleton.

EXPERIMENTAL

General. All reagents and solvents were used without further purification. Melting points were determined with a Gallenkamp capillary melting point apparatus and are uncorrected. ¹H NMR spectra were recorded on a Hitachi R-1500 (FT, 60 MHz) spectrometer. Chemical shifts are expressed in δ units relative to tetramethylsilane as internal standard. IR spectra were recorded by using on a JASCO FT/IR-300E spectrometer. Mass spectral data were obtained on a Hewlett Packard 5970 GC/MS system. Silica gel 60 (70–230 mesh, E. Merck) was used for all column chromatographic separations.

ω -(Methylsulfinyl)acetophenone (1**).** A solution of sodium metaperiodate (5.14 g, 24 mmol) in water (30 mL) was added in small portions to a stirred solution of 2-(methylthio)acetophenone (4 g, 24 mmol) in methanol (60 mL) at 0 °C and the mixture was further stirred at room temperature for 12 h. Inorganic materials were filtered off and the filtrate was extracted with chloroform (3×30 mL). The combined organic layer was dried over MgSO₄, and concentrated under reduced pressure. The residual solid was recrystallized from ethyl acetate to give **1** in 80% yield (3.49 g). mp. 86–87 °C (lit.⁸ 86–86.5 °C); IR (KBr) 3044, 2933, 1675 (C=O), 1577, 1422, 1299, 1193, 1030 (S=O), 978 cm⁻¹; ¹H NMR (CDCl₃) δ 2.77 (s, 3H), 4.39 (d, J=3.5Hz, 2H), 7.48–7.62 (m, 3H), 7.98 (d, J=7.6Hz, 2H).

General procedure for the synthesis of 3-methylthio-2-phenylbenzofuran (2**).** A solution of **1** (1.1 mmol), substituted phenol (1.1 mmol), and anhydrous *p*-toluenesulfonic acid (3.3 mmol) in 1,2-dichloroethane (15 mL) was refluxed for 1 h. Then the mixture was cooled at room temperature, washed with water to remove *p*-toluenesulfonic acid, and dried over MgSO₄. The solvent was evaporated off, and the residue was purified by column chromatography (hexane/ethyl acetate=6/1) to give

2. 2a: Yield 80%, mp 67-68 °C; ^1H NMR (CDCl_3) δ 2.37 (s, 3H), 2.48 (s, 3H), 6.80-8.37 (m, 8H). **2b:** Yield 79%, mp 37-38 °C; ^1H NMR (CDCl_3) δ 1.31 (t, $J=7.6\text{Hz}$, 3H), 2.38 (s, 3H), 2.80 (q, $J=7.6\text{Hz}$, 2H), 7.06-8.37 (m, 8H). **2c:** Yield 82%, colorless liquid, ^1H NMR (CDCl_3) δ 1.33 (d, $J=6.5\text{Hz}$, 6H), 2.38 (s, 3H), 2.84-3.31 (m, 1H), 7.10-8.36 (m, 8H). **2d:** Yield 83%, colorless liquid, ^1H NMR (CDCl_3) δ 1.42 (s, 9H), 2.39 (s, 3H), 7.24-8.38 (m, 8H). **2e:** Yield 78%, mp 115-116 °C; ^1H NMR (CDCl_3) δ 2.38 (s, 3H), 2.47 (s, 3H), 2.53 (s, 3H), 6.98-8.33 (m, 7H). **2f:** Yield 53%, colorless liquid, ^1H NMR (CDCl_3) δ 2.32 (s, 3H), 2.42 (s, 3H), 2.49 (s, 3H), 6.86-8.22 (m, 7H). The above spectral data for **2** are in accord with those reported.^{6b}

ω -Methylsulfinyl-*p*-methylacetophenone (3). A suspension of NaH (60% mineral oil dispersion, 2 g, 50 mmol) in DMSO (30 mL) was heated with stirring at 70-75 °C for 40 min under Ar. After cooling to room temperature, THF (15 mL) was added to the reaction mixture. Ethyl *p*-toluate (3.28 g, 20 mmol) was added to the mixture at 0 °C, and the stirring was continued for 90 min at the room temperature. The reaction mixture was poured into water (150 mL), acidified with aqueous HCl to a pH 3~4, and thoroughly extracted with chloroform (3 \times 30 mL). The combined organic layer was washed with water (3 \times 30 mL), dried over MgSO_4 , and evaporated off. The residual solid was recrystallized from ethyl acetate to give **3** in 74% yield (2.9 g). mp. 113-114 °C; IR (KBr) 2998, 2911, 2362, 2344, 1671 ($\text{C}=\text{O}$), 1605, 1560, 1411, 1278, 1186, 1131, 1044 ($\text{S}=\text{O}$), 961 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.44 (s, 3H), 2.75 (s, 3H), 4.33 (d, $J=4.1\text{Hz}$, 2H), 7.31 (d, $J=8.2\text{Hz}$, 2H), 7.89 (d, $J=8.2\text{Hz}$, 2H); MS m/z 196 (M^+).

ω -Methylsulfinyl-*p*-chloroacetophenone (4). By the same procedure as described above for the preparation of **3**, compound **4** was obtained from ethyl *p*-chlorobenzoate (3.69 g, 20 mmol), NaH (60% mineral oil dispersion, 2 g, 50 mmol), and DMSO (30 mL) in 79% yield (3.42 g). mp 126-127 °C; IR (KBr) 3033, 2988, 2921, 2377, 2344, 1666 ($\text{C}=\text{O}$), 1588 1422, 1288, 1033 ($\text{S}=\text{O}$), 767 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.76 (s, 3H), 4.36 (d, $J=3.9\text{Hz}$, 2H), 7.49 (d, $J=8.2\text{Hz}$, 2H), 7.94 (d, $J=8.2\text{Hz}$, 2H); MS m/z 216 (M^+).

General procedure for the synthesis of 3-methylthio-2-(*p*-tolyl) benzofuran (5). By the same procedure as

described above for the preparation of **2**, compound **5** was obtained from **3** (2 mmol), substituted phenol (2 mmol), and anhydrous *p*-toluenesulfonic acid (6 mmol). **5a:** Yield 78%, mp 97-98 °C; IR (KBr) 2918, 1499, 1472, 1332, 1272, 1256, 1202, 1187, 1078, 1016, 970 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.36 (s, 3H), 2.40 (s, 3H), 2.48 (s, 3H), 7.11-7.57 (m, 5H), 8.17 (d, $J=8.2\text{Hz}$, 2H); MS m/z 268 (M^+). **5b:** Yield 83%, mp 66-67 °C; IR (KBr) 2959, 2918, 1498, 1468, 1412, 1275, 1255, 1202, 1185, 1081, 1021, 969 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.31 (t, $J=7.6\text{Hz}$, 3H), 2.37 (s, 3H), 2.41 (s, 3H), 2.79 (q, $J=7.6\text{Hz}$, 2H), 7.11-7.54 (m, 5H), 8.17 (d, $J=8.2\text{Hz}$, 2H); MS m/z 282 (M^+). **5c:** Yield 82%, colorless liquid, IR (neat) 2952, 2920, 1613, 1501, 1470, 1420, 1255, 1204, 1184, 1078, 966 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.33 (d, $J=7.0\text{Hz}$, 6H), 2.38 (s, 6H), 2.82-3.38 (m, 1H), 7.02-7.54 (m, 5H), 8.17 (d, $J=8.2\text{Hz}$, 2H); MS m/z 296 (M^+). **5d:** Yield 86%, colorless liquid, IR (neat) 2916, 2920, 2867, 1501, 1471, 1363, 1332, 1278, 1259, 1205, 1185, 1077, 1019, 969 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.42 (s, 9H), 2.37 (s, 6H), 7.11-7.69 (m, 5H), 8.17 (d, $J=8.2\text{Hz}$, 2H); MS m/z 310 (M^+). **5e:** Yield 67%, mp 125-126 °C; IR (KBr) 2917, 1497, 1456, 1442, 1254, 1120, 1187, 1076, 1065, 1012, 971 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.36 (s, 3H), 2.42 (s, 3H), 7.15-7.68 (m, 5H), 8.17 (d, $J=8.2\text{Hz}$, 2H). MS m/z 288 (M^+).

General procedure for the synthesis of 3-methylthio-2-(*p*-chlorophenyl)benzofuran (6). By the same procedure as described above for the preparation of **2**, compound **6** was obtained from **4** (2 mmol), substituted phenol (2 mmol), and anhydrous *p*-toluenesulfonic acid (6 mmol). **6a:** Yield 87%, mp 107-108 °C; IR (KBr) 2920, 1486, 1470, 1401, 1254, 1202, 1090, 1074, 1010, 971 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.37 (s, 3H), 2.49 (s, 3H), 7.07-7.52 (m, 5H), 8.26 (d, $J=8.7\text{Hz}$, 2H); MS m/z 288 (M^+). **6b:** Yield 90%, mp 64-65 °C; IR (KBr) 2955, 2925, 2865, 1544, 1485, 1467, 1400, 1337, 1254, 1202, 1092, 1076, 1009, 966 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.31 (t, $J=7.6\text{Hz}$, 3H), 2.37 (s, 3H), 2.79 (q, $J=7.6\text{Hz}$, 2H), 7.11-7.52 (m, 5H), 8.25 (d, $J=8.7\text{Hz}$, 2H); MS m/z 302 (M^+). **6c:** Yield 84%, mp 81-82 °C; IR (KBr) 2952, 2919, 2862, 1545, 1486, 1463, 1422, 1402, 1383, 1254, 1204, 1177, 1094, 1077, 1012, 969 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.33 (d, $J=7.0\text{Hz}$, 6H), 2.38 (s, 3H), 2.71-3.19 (m, 1H), 7.10-5.53 (m, 5H), 8.25 (d, $J=8.7\text{Hz}$, 2H); MS m/z 316 (M^+). **6d:** Yield 90%, mp 83-84 °C; IR (KBr) 2959,

1472, 1400, 1360, 1273, 1257, 1206, 1090, 1076, 1028, 1011 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.42 (s, 9H), 2.38 (s, 3H), 7.16-7.71 (m, 5H), 8.25 (d, $J=8.2\text{Hz}$, 2H); MS m/z 330 (M^+). **6e**: Yield 75%, mp 136-137 $^{\circ}\text{C}$; IR (KBr) 2942, 2929, 1483, 1456, 1442, 1400, 1253, 1198, 1092, 1065, 1009, 971 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.37 (s, 3H), 7.18-7.69 (m, 5H), 8.25 (d, $J=8.2\text{Hz}$, 2H); MS m/z 308 (M^+).

General procedure for the synthesis of 2-(*p*-tolyl)benzofuran (7). Compound **5** (1.2 mmol) was heated at 60-65 $^{\circ}\text{C}$ in ethanol (30 mL) containing Raney nickel (W-2, 1.8 g) for 1h. The Raney nickel was removed by filtration and the solvent was evaporated off. The residual solid was recrystallized from ethanol to give **7. 7a**: Yield 95%, mp 154-155 $^{\circ}\text{C}$; IR (KBr) 2962, 1587, 1505, 1467, 1290, 1266, 1207, 1121, 1036, 1015, 912 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.39 (s, 3H), 2.43 (s, 3H), 6.89 (s, 1H), 6.9 7-7.47 (m, 5H), 7.75 (d, $J=8.2\text{Hz}$, 2H); MS m/z 222 (M^+). **7b**: Yield 95%, mp 108-109 $^{\circ}\text{C}$; IR (KBr) 2911, 1505, 1463, 1289, 1265, 1209, 1195, 1111, 1037, 1015, 932 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.28 (t, $J=7.6\text{Hz}$, 3H), 2.38 (s, 3H), 2.74 (q, $J=7.6\text{Hz}$, 2H), 6.89 (s, 1H), 7.01-7.51 (m, 5H), 7.74 (d, $J=8.2\text{Hz}$, 2H); MS m/z 236 (M^+). **7c**: Yield 95%, mp 97-98 $^{\circ}\text{C}$; IR (KBr) 2956, 2866, 1506, 1471, 1381, 1354, 1288, 1264, 1208, 1160, 1112, 1034, 1015, 916 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.18 (d, $J=7.0\text{Hz}$, 6H), 2.38 (s, 3H), 2.68-3.23 (m, 1H), 6.90 (s, 1H), 7.03-7.50 (m, 5H), 7.74 (d, $J=8.2\text{Hz}$, 2H); MS m/z 250 (M^+). **7d**: Yield 95%, mp 124-125 $^{\circ}\text{C}$; IR (KBr) 2956, 1505, 1473, 1458, 1364, 1328, 1277, 1208, 1166, 1127, 1036, 1016, 913 cm^{-1} ; ^1H NMR (CDCl_3) 1.39 (s, 9H), 2.38 (s, 3H), 6.92 (s, 1H), 7.06-7.66 (m, 5H), 7.75 (d, $J=8.2\text{Hz}$, 2H); MS m/z 264 (M^+). **7e**: Yield 95%, mp 177-178 $^{\circ}\text{C}$; IR (KBr) 2955, 1585, 1504, 1444, 1262, 1206, 1163, 1114, 1060, 1035, 1015, 927 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.39 (s, 9H), 6.94 (s, 1H), 7.01-7.58 (m, 5H), 7.74 (d, $J=8.2\text{Hz}$, 2H); MS m/z 242 (M^+).

General procedure for the synthesis of 2-(*p*-chlorophenyl) benzofuran (8). By the same procedure as described above for the preparation of **7**, compound **8** was obtained from **6** (1 mmol), Raney nickel (1.5 g), and ethanol (25 mL). The residual solid was recrystallized from ethanol to give **8. 8a**: Yield 88%, mp 186-187 $^{\circ}\text{C}$;

IR (KBr) 2913, 1579, 1488, 1462, 1404, 1261, 1210, 1195, 1092, 1034, 1009, 914 cm^{-1} ; ^1H NMR (CDCl_3) δ 2.43 (s, 3H), 6.91 (s, 1H), 6.99-7.55 (m, 5H), 7.77 (d, $J=8.8\text{Hz}$, 2H); MS m/z 242 (M^+). **8b**: Yield 84%, mp 149-150 $^{\circ}\text{C}$; IR (KBr) 2970, 1487, 1466, 1403, 1264, 1193, 1089, 1033, 1009, 911 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.29 (t, $J=7.0\text{Hz}$, 3H), 2.75 (q, $J=7.6\text{Hz}$, 2H), 6.94 (s, 1H), 7.01-7.50 (m, 5H), 7.78 (d, $J=8.2\text{Hz}$, 2H); MS m/z 256 (M^+). **8c**: Yield 89%, mp 152-153 $^{\circ}\text{C}$; IR (KBr) 2958, 1580, 1488, 1469, 1403, 1355, 1267, 1160, 1117, 1103, 1091, 1031, 1009, 915 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.30 (d, $J=7.0\text{Hz}$, 6H), 2.80-3.23 (m, 1H), 6.94 (s, 1H), 7.09-7.56 (m, 5H), 7.78 (d, $J=8.8\text{Hz}$, 2H); MS m/z 270 (M^+). **8d**: Yield 90%, mp 154-155 $^{\circ}\text{C}$; IR (KBr) 2956, 1579, 1471, 1402, 1366, 1328, 1275, 1166, 1127, 1088, 1032, 1009, 913 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.39 (s, 9H), 6.96 (s, 1H), 7.10-7.57 (m, 5H), 7.78 (d, $J=8.8\text{Hz}$, 2H); MS m/z 284 (M^+). **8e**: Yield 76%, mp 151-152 $^{\circ}\text{C}$; IR (KBr) 2957, 1599, 1581, 1486, 1443, 1324, 1273, 1260, 1163, 1091, 1061, 1033, 1011, 926 cm^{-1} ; ^1H NMR (CDCl_3) δ 6.93 (s, 1H), 7.15-7.86 (m, 7H); MS m/z 262 (M^+).

REFERENCES

1. Talapatra, B.; Ray, T.; Talapatra, S. K. *J. Indian Chem. Soc.* **1978**, 55, 1204.
2. Levan, N.; Pham, T. V. *Phytochemistry* **1981**, 20, 485.
3. Bohlmann, F.; Ahmed, M.; Robinson, H.; King, R. M. *Phytochemistry* **1981**, 20, 1493.
4. Achenbach, H.; Gross, J.; Dominguez, X. A.; Star, J. V.; Salgado, F. *Phytochemistry* **1987**, 26, 2041.
5. Sheen, W. S.; Tsai, I. L.; Teng, C. M.; Chen, I. S. *Phytochemistry* **1994**, 36, 213.
6. (a) Choi, H. D.; Seo, P. J.; Son, B. W. *J. Korean Chem. Soc.* **1999**, 43, 237. (b) Choi, H. D.; Seo, P. J.; Son, B. W. *J. Korean Chem. Soc.* **1999**, 43, 606. (c) Seo, P. J.; Ha, M. C.; Choi, H. D.; Son, B. W. *J. Korean Chem. Soc.* **2000**, 44, 391.
7. Choi, H. D.; Seo, P. J. *J. Korean Chem. Soc.* **2001**, 45, 274.
8. Corey, E. J.; Chaykovsky, M. *J. Am. Chem. Soc.* **1965**, 87, 1345.
9. Schneiders, G. E.; Stevenson, R. *J. Org. Chem.* **1979**, 44, 4710.