

Supplementary Material

Energy- and Time-Dependent Branching to Competing Paths in Coupled Unimolecular Dissociations of Chlorotoluene Radical Cations[†]

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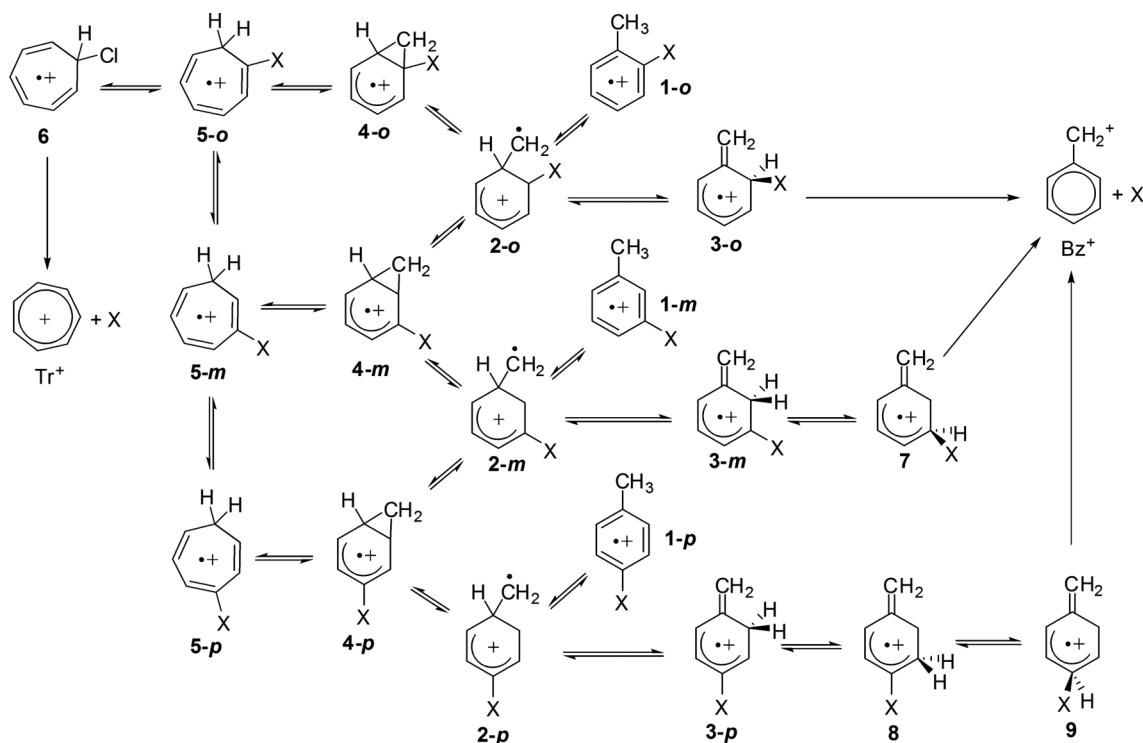


Figure S1. The proposed mechanism for the unimolecular dissociation of the *o*-, *m*-, and *p*-halotoluene radical cations ($C_7H_7X^+$) denoted by **1-x** ($x = o, m$, and p). The benzylium and tropylium product ions are denoted by Bz^+ and Tr^+ , respectively. $X = Cl$, in the present study.

[†]This paper is to commemorate Professor Myung Soo Kim's honourable retirement.

Table S1. Relative Energies of the Local Minima and the Transition States on the Dissociation Pathways of *o*-, *m*-, and *p*-Chlorotoluene Radical Cations at 0 K^a

<i>o</i> -isomer	<i>m</i> -isomer		<i>p</i> -isomer		
transient species	relative energy ΔE (kcal mol ⁻¹)	transient species	relative energy ΔE (kcal mol ⁻¹)	transient species	relative energy ΔE (kcal mol ⁻¹)
1-o	0.0	1-m	0.0	1-p	0.0
<i>o</i> -tolyl + Cl	75	<i>m</i> -tolyl + Cl	77	<i>p</i> -tolyl + Cl	82
TS1	38	TS6	41	TS12	40
2-o	36	2-m	40	2-p	38
benzylum dissociation channel					
2-o	36	2-m	40	2-p	38
TS2	41	TS7	41	TS13	37
3-o	18	3-m	7.3	3-p	14
TSB-o	32	TS9	37	TS15	34
Bz ⁺ + Cl	34	7	33	8	24
		TSB-m	33	TS16	36
		Bz ⁺ + Cl	34	9	25
				TSB-p	36
				Bz ⁺ + Cl	37
tropylium dissociation channel					
2-o	36	2-m	40	2-p	38
TS3	38	TS8	43	TS14	38
4-o	30	4-m	21	4-p	27
TS5	47	TS10	34	TS17	33
6	23	5-m	20	5-p	20
TST	26	TS11	45	TS18	45
Tr ⁺ + Cl	26	5-o	16	5-m	22
		TS5	47	TS11	48
		6	23	5-o	18
		TST	26	TS5	50
		Tr ⁺ + Cl	26	6	26
				TST	29
				Tr ⁺ + Cl	29
coupling processes					
2-o	36	2-m	40	4-p	27
TSOM	37	TSMP	42	TSMP	45
4-m	20	4-p	24	2-m	43
		4-m	21		
		TSOM	38		
		2-o	37		

^aRelative energies are with respect to the reactant **1-x** (*x* = *o*, *m*, and *p*)

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