

Supporting Information

Lipase Catalyzed Kinetic Resolution of *rac*-2-(3-Methoxy-4-methylphenyl) propan-1-ol and *rac*-2-(3-Hydroxy-4-methylphenyl)propyl propanoate for *S*-(+)-XanthorrhizolAzam Sharif Mohammed Shafioul^{†,‡} and Chan Seong Cheong^{†,*}[†]University of Science & Technology, Daejeon, South Korea[‡]Biomolecules Function Research Center, Future Convergence Research Division, Korea Institute of Science and Technology, Seoul 136-791, Korea. *E-mail: c2496@kist.re.kr

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Table 1. Conversions, enantiomeric excess and enantioselectivity of enzyme catalyzed hydrolysis reactions of *rac*-substrate V^a

Enzyme	Time (h)	<i>c</i> (%) ^c	ee _s (%) ^b	ee _p (%) ^b	<i>E</i> ^c
Cal B	4	4	3	64	6
PLE	6	29	5	13	1
CRL	4	47	16	18	2
LAN	13	40	20	31	2
Acylase	13	64	23	13	2
LAK	5	49	17	18	2
PCL	4	75	25	8	1
LAH	3	69	29	13	2
AOL	19	53	99	87	80
PPL	19	67	99	50	14

^aReaction condition: 10 mg *rac*-substrate V, enzyme 30 mg, a mixture of acetone: phosphate buffer (pH 7.0) = 1:4 was shaken at 30 °C and 250 rpm. Here *c*: conversion, *E*: enantioselectivity, ee_s: enantiomeric excess of starting material, ee_p: enantiomeric excess of product. ^bDetermined by HPLC using (R,R)-Whelk-01, 250 × 4.6 mm, 5 μm column, mobile phase: hexanes: ethanol: diethyl amine = 97:03:0.05 % v/v with flow rate 0.4 mL/min at RT. ^cDetermined from the equation described in 2.7.

Table 2. Conversion, enantiomeric excess and enantioselectivity for enzyme catalyzed transesterification^a and hydrolysis^b reaction

Enzyme	Time (h)	Wt. (Substrate)	<i>c</i> (%) ^c	ee _s (%)	ee _p (%)	<i>E</i> ^c
AOL	34	10 g (II) ^a	58	98 ^c	72 ^c	27 ± 1
AOL	23	1 g (V) ^b	53	99 ^d	88 ^d	80 ± 3
Bi-phase reaction (hexanes : phosphate buffer pH 7.0 = 2:4)						
AOL	48	0.1 g (V) ^b	58	99 ^d	72 ^d	31 ± 3

^aReaction condition: *rac*-substrate II, vinyl propanoate (2.5 eq.), AOL Enzyme (3 mass equivalent), *tert*-BuOMe was shaken at 25 °C and 250 rpm.

^bReaction condition: *rac*-substrate V, AOL Enzyme (3 mass eq.), a mixture of acetone: phosphate buffer (pH 7.0) = 1:4 was shaken at 30 °C and 250 rpm. Here *c*: conversion, *E*: enantioselectivity, ee_s: enantiomeric excess of starting material, ee_p: enantiomeric excess of product. ^cDetermined by HPLC using Chiralpak AS-H, 250 × 4.6 mm, 5 μm column, mobile phase: hexanes: IPA = 95:05% v/v with flow rate 0.4 mL/min at RT. ^dDetermined by HPLC using (R,R)-Whelk-01, 250 × 4.6 mm, 5 μm column, mobile Phase: hexanes: ethanol: diethyl amine = 97:03:0.05% v/v with flow rate 0.4 mL/min at rt. Values of *c*, ee_s and ee_p are the average of four determinations. ^eDetermined from the equation described in 2.7 with average of *c* and ee_s.

Table 3. Solubility of the methyl protected and deprotected alcohol and propanoate ester

<i>rac</i> -compound	Solubility (g/L) ^a	Type
II	2.5	Slightly soluble
III	7.1	Slightly soluble
IV	0.052	Partially soluble
V	0.089	Partially soluble

^aSolubility was measured with Nephelostar instrument from BMG Lab Tech. Values are the average of three determinations.