

Supplementary Materials

Kinetics and Mechanism of the Anilinolysis of Diisopropyl Thiophosphinic Chloride in Acetonitrile

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Received August 18, 2011, Accepted August 31, 2011

Substrate : Diisopropyl thiophosphinic chloride

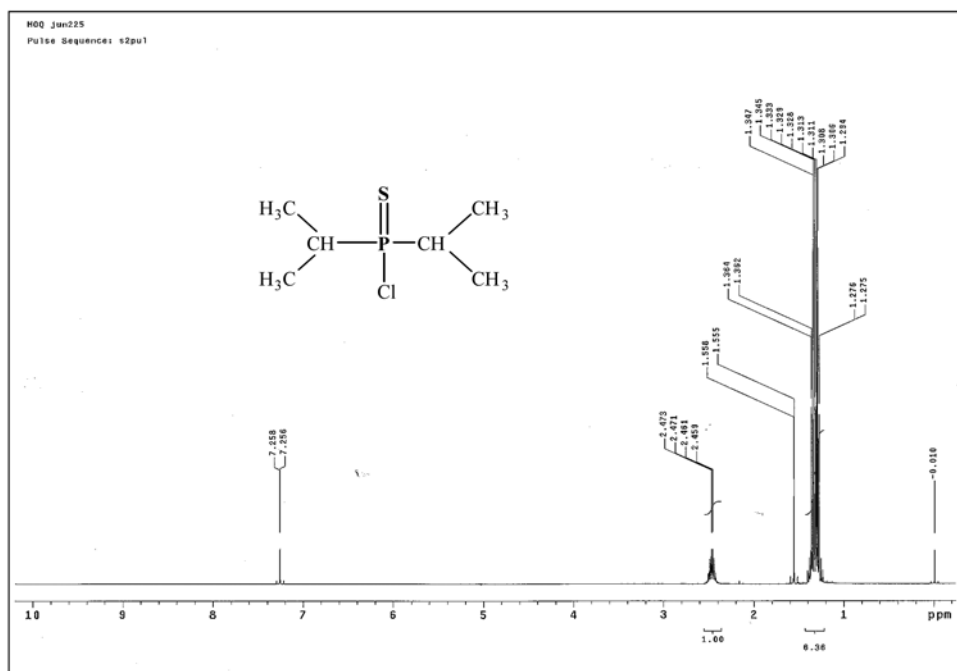
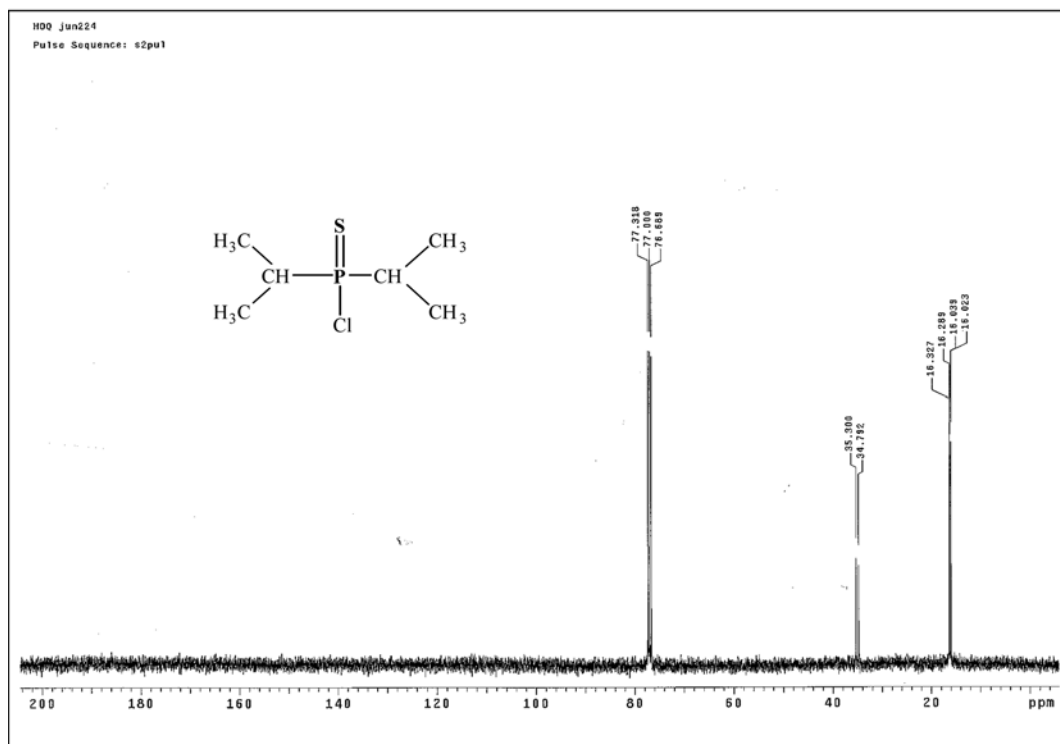
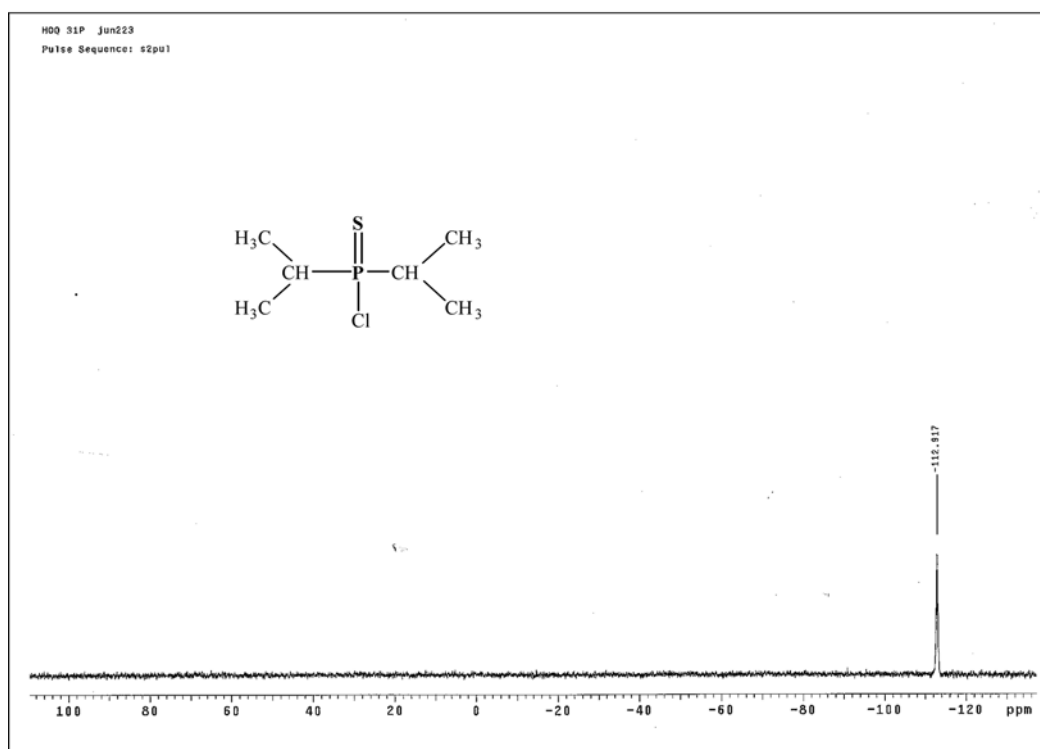


Figure S1. The ¹H-NMR spectrum of diisopropyl thiophosphinic chloride

Substrate : Diisopropyl thiophosphinic chloride**Figure S2.** The ¹³C-NMR spectrum of diisopropyl thiophosphinic chloride**Figure S3.** The ³¹P-NMR spectrum of diisopropyl thiophosphinic chloride

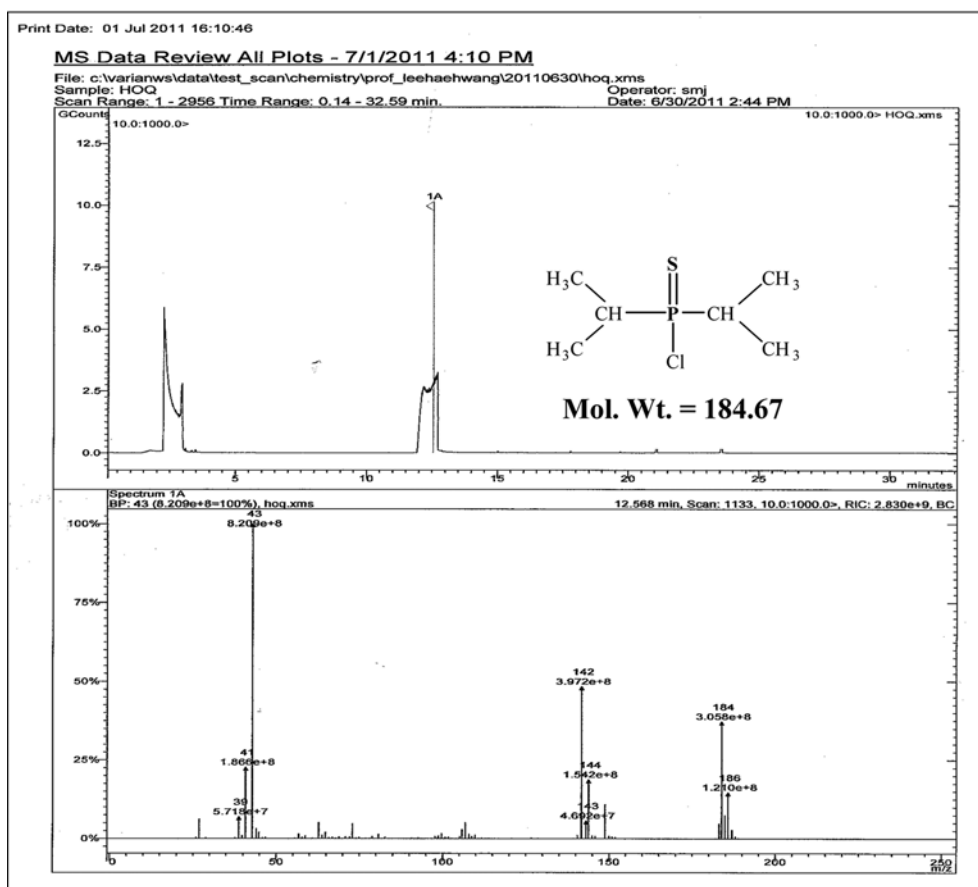
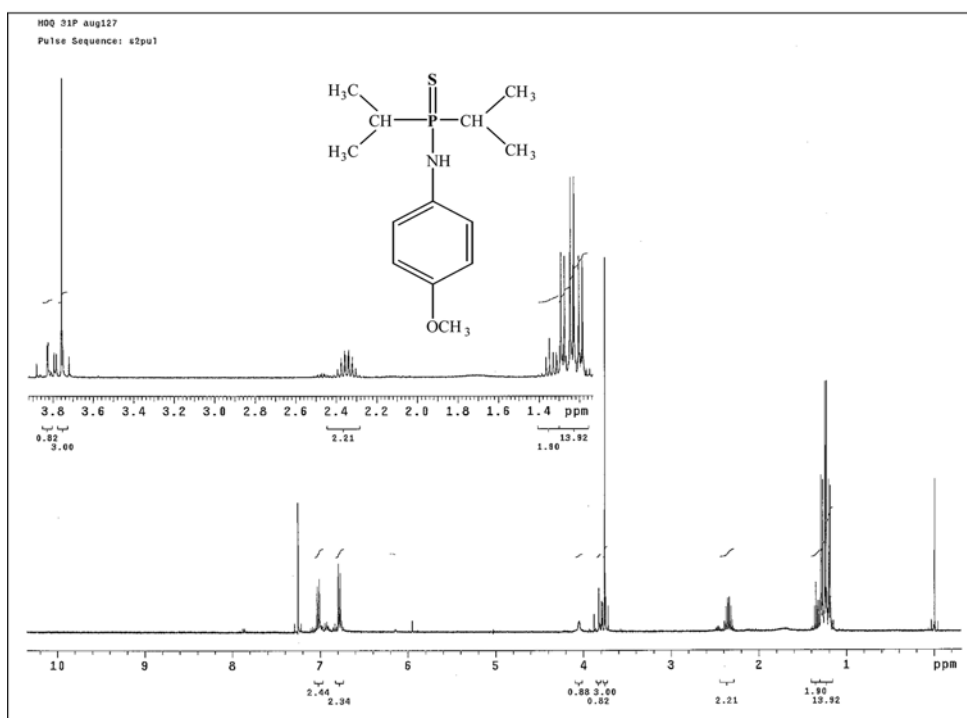


Figure S4. The GC-MS spectrum of diisopropyl thiophosphinic chloride

Product : [(i-Pr)₂P(=S)NHC₆H₄(4-OCH₃)]

Figure S5. The ¹H-NMR spectrum of [(i-Pr)₂P(=S)NHC₆H₄(4-OCH₃)]

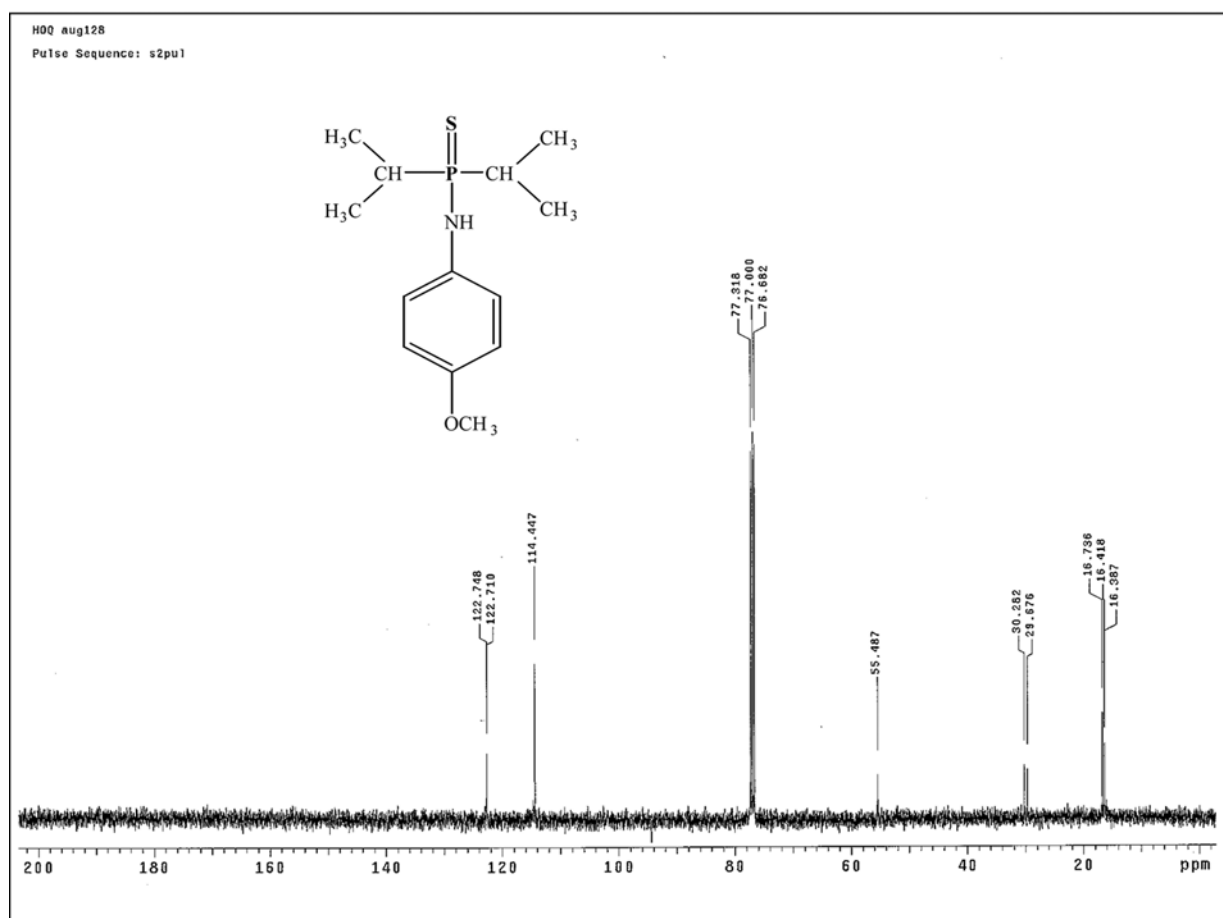


Figure S6. The ^{13}C -NMR spectrum of $[(i\text{-Pr})_2\text{P}(=\text{S})\text{NHC}_6\text{H}_4(4\text{-OCH}_3)]$

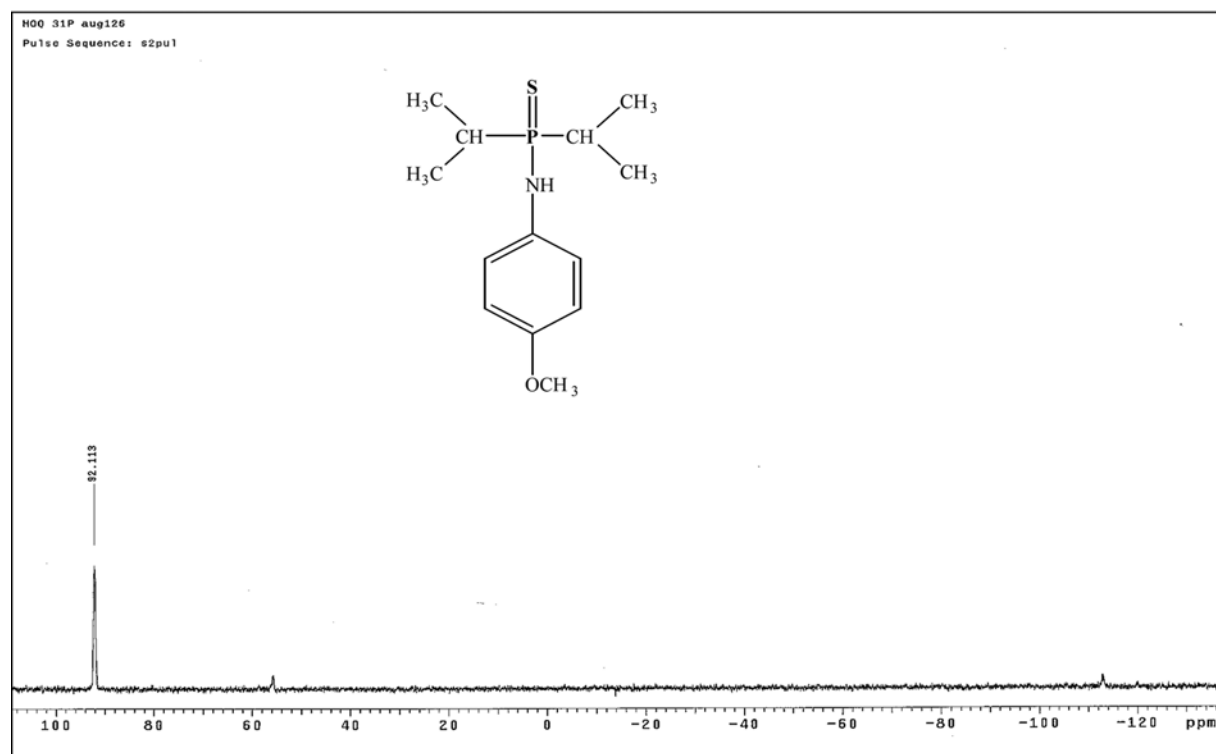


Figure S7. The ^{31}P -NMR spectrum of $[(i\text{-Pr})_2\text{P}(=\text{S})\text{NHC}_6\text{H}_4(4\text{-OCH}_3)]$

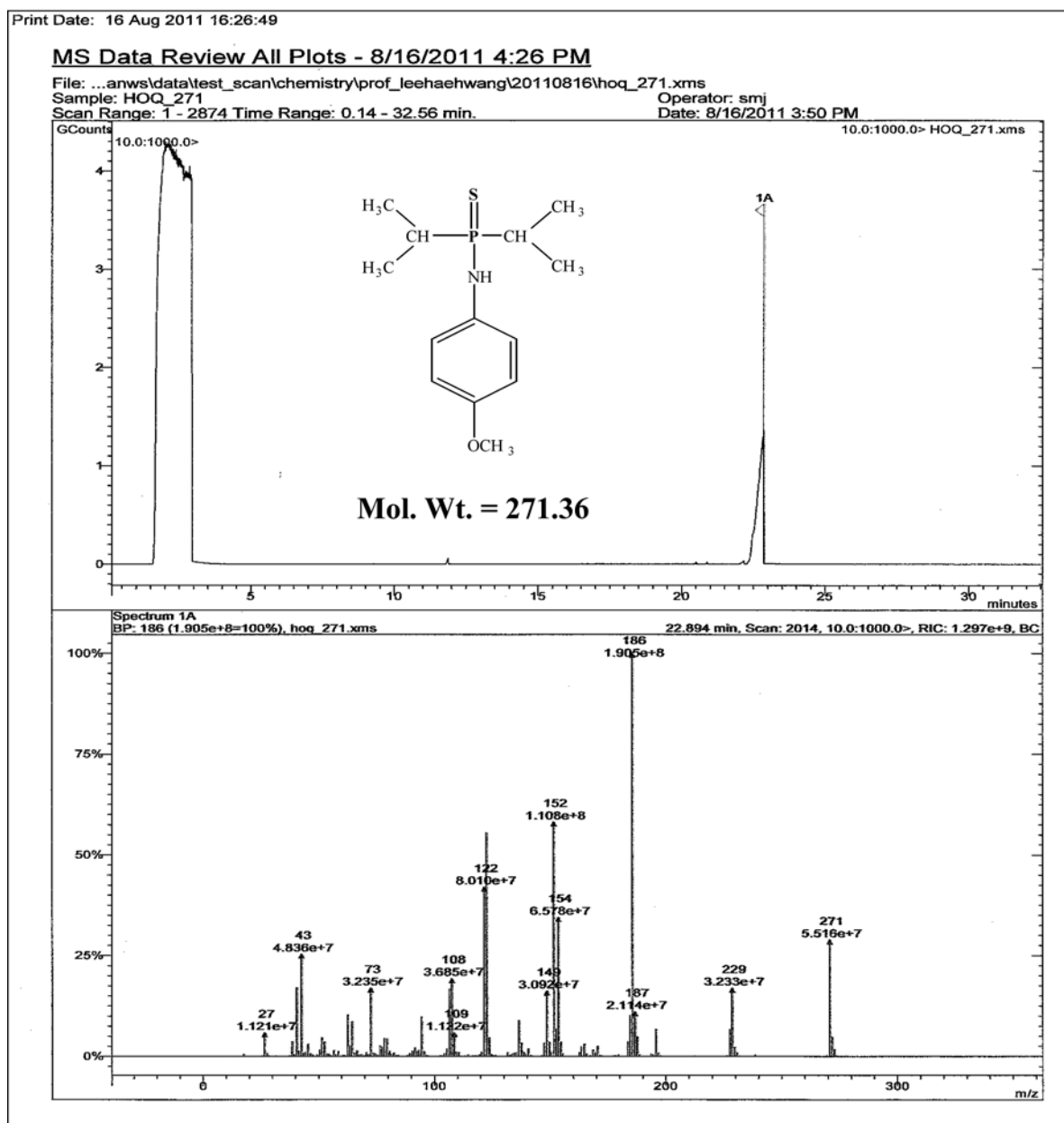


Figure S8. The GC-MS spectrum of [(i-Pr)₂P(=S)NHC₆H₄(4-OCH₃)]