

Synthesis of Halicyclindramide A Mimetics Containing Lactone Isosteres[†]

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Synthesis of halicyclindramide A mimetics *via* solid-phase peptide synthesis is described. The *N*-Me-Gly-Thr residues in halicyclindramide A has been replaced by *N*-Me-Gly-Dpr, Lys, or *N*-Me-Gly-*allo*-Thr. Solution structures of the mimetics were compared by CD spectroscopy in an aqueous buffer and in TFE.

Key Words : Synthesis, Halicyclindramide A, Mimetic, Depsipeptide

Introduction

Halicyclindramides A-D¹ are cyclodepsipeptides of marine origin with potent antifungal and cytotoxic activities. These molecules are amphiphilic and share many unusual amino acids, such as D-amino acids, *N*-methyl amino acids, and cysteic acid, and contain a formyl-D-Ala residue at the *N* terminus (Figure 1). Other cyclic depsipeptides, such as discodermins A-H,² polydiscamide A-D,^{2b,3} microspinosamide,^{2c} and corticiamide A⁴ have also been isolated from marine sponges, and they possess unusual amino acids such as those found in halicyclindramides. Notably, the presence of D-cysteic acid is a unique characteristic, shared only by these depsipeptides.

These peptides possess various activities, such as antimicrobial, anticancer, and anti-HIV-1 activities, and are known to interact with phospholipase A₂ or G protein-coupled receptor (GPCR). Although the conformations and modes of action of these peptides have yet to be elucidated, their activities appear to be related to their association with membrane proteins as a result of their amphiphilic character. Increased membrane permeabilization of Ca²⁺ and ATP by

discodermin A was reported in 2001.⁵

In 2009, we reported the first synthesis of halicyclindramide A and its conformational change by a CD study in a membrane environment.⁶ However, no other depsipeptides containing D-cysteic acid have been synthesized to date. In this paper, we describe a solid-phase synthesis of halicyclindramide A mimetics where the *N*-Me-Gly-Thr, the ester linkage of the cyclic part, is replaced by *N*-Me-Gly-Dpr, Lys, or *N*-Me-Gly-*allo*-Thr (Figure 1). A preliminary conformational study by CD spectroscopy was also performed.

Conformational restriction imposed by ring formation is known to be important for activity and stability and is often adopted in peptidomimetic studies. Syntheses of halicyclindramide mimetics containing lactone isosteres would therefore be valuable in exploring halicyclindramide conformational changes and in understanding the mechanism of the membrane interactions of these peptides.

Results and Discussion

For the syntheses of mimetics of halicyclindramide A, we applied the same strategy as we used for the synthesis of halicyclindramide A.⁶ So, Rink amide resin was anchored to the side chain of the Asp residue and macrolactamization was performed on the resin. After coupling of all the amino acids, oxidation of the resin-bound Cys residue was carried out and, after the resin cleavage, the last step was formylation of the *N*-terminal Ala residue.

The resin-bound peptide **2** was prepared as in the previous synthesis of halicyclindramide A (Scheme 1). To synthesize the amide mimetic **1a**, we incorporated Fmoc-Dpr(Mtt)-OH to provide **3**. The side chain of the amino acid Dpr should be protected with a 4-methyltrityl (Mtt) group, which can be selectively removed under mild acidic conditions to enable coupling of further amino acids. The Mtt group of the Dpr residue was removed under 1% TFA, and then a free amino group was coupled to Alloc-Sar-OH to yield the resin-bound peptide **4**. The Alloc and allyl protecting groups at the Sar and Asn residues were removed selectively by treatment with Pd(Ph₃P)₄.⁷ The deprotected peptide was then cyclized between the Sar and Asn residues with an excess of a HATU/HOAt/NMM mixture, providing the resin-bound cyclic peptide

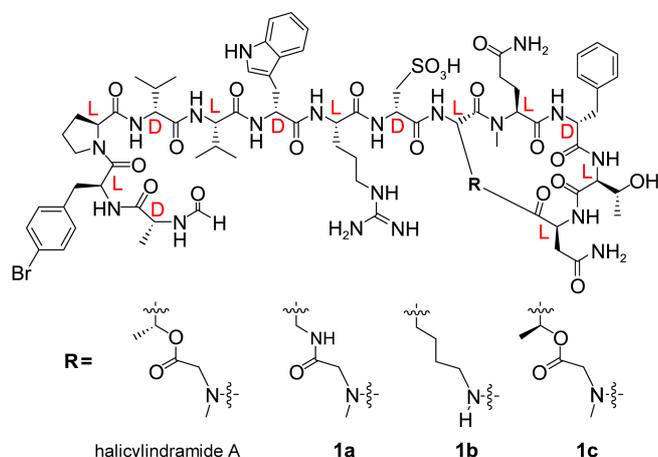
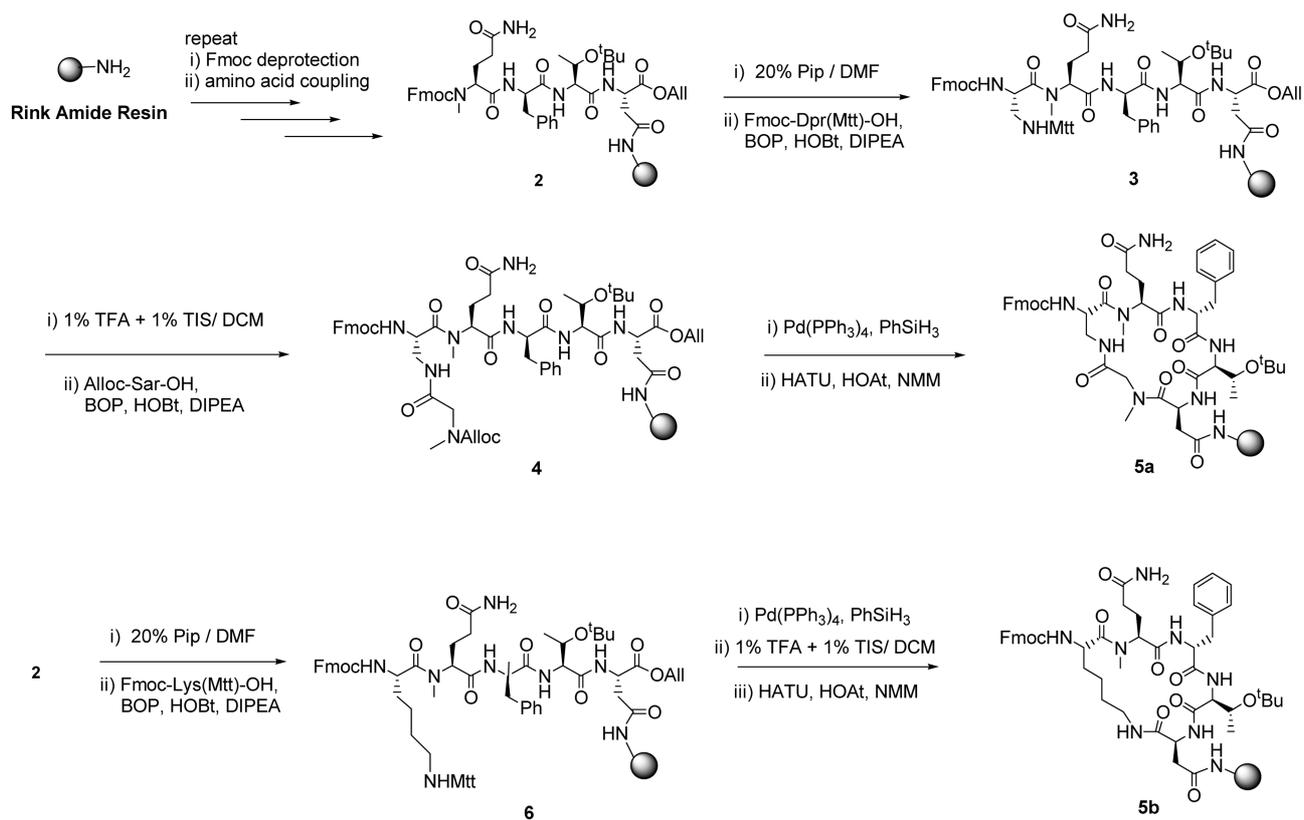
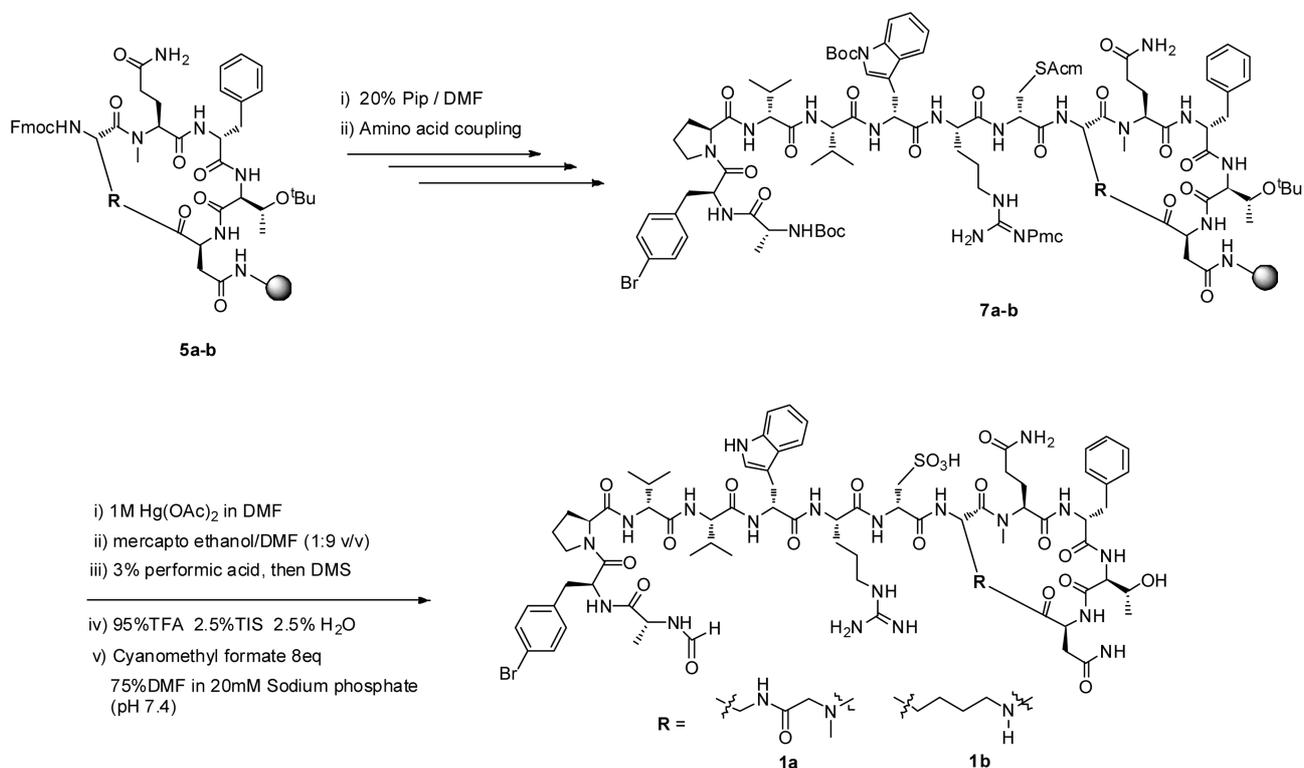


Figure 1. Structure of halicyclindramide A and its mimetics.

[†]This paper is dedicated to Professor Eun Lee on the occasion of his honourable retirement.


 Scheme 1. Synthesis of **5a** and **5b**.

 Scheme 2. Synthesis of **1a-b**.

5a. The cyclization reaction was assessed using the chloranil test to detect the absence of the secondary amine of the Sar residue.⁸

For the preparation of **5b**, an alkane mimetic of halicyclindramide A, Fmoc-Lys(Mtt)-OH was used to provide the peptide **6**. The allyl and Mtt protecting groups were removed selectively by employing the same deprotection conditions as for the synthesis of **5a**, and the resulting peptide was then cyclized to obtain the resin-bound peptide **6a**.

The cyclic peptides **5a-b** were further elongated one by one with the remaining *N*-terminal amino acids, using PyBOP/HOAt/DIPEA to afford compounds **7a-b** (Scheme 2). Coupling of the D-Cys analog was performed while the side chain of the Cys residue was protected with an acetamidomethyl (Acm) group, which can be removed selectively and oxidized to cysteic acid. When we attempted segment coupling of the *N*-terminal peptide onto **5a**, no coupled product could be obtained after resin cleavage.

The indole ring of the Trp residue can be easily oxidized to form various byproducts under oxidative conditions.⁹ Hence, the Cys residue was oxidized while the Trp residue was protected with a Boc group in the solid phase. The resin was treated with 1 M Hg(OAc)₂¹⁰ in DMF to deprotect the Acm group selectively from the Cys residue and washed with 10% mercaptoethanol to remove Hg²⁺ ions (Scheme 2). The free thiol was then oxidized with freshly prepared 3% performic acid¹¹ and the resin was treated with dimethylsulfide

to remove excess oxidant. The oxidized cyclic peptide from the performic acid oxidation was cleaved from the resin and purified by RP-HPLC to provide two deformyl mimetics of halicyclindramide A.

Finally, the *N*-terminal Ala residue was formylated using an excess of cyanomethyl formate¹² in a solution of DMF-sodium phosphate buffer (pH 7.4) at room temperature. The crude product was purified by semipreparative RP-HPLC to give **1a-b**. The pure peptides **1a-b** were analyzed by ¹H and 2D NMR, HRMS, CD, and analytical RP-HPLC.

For the synthesis of **1c**, an *allo*-Thr derivative of halicyclindramide A, we followed the same strategy for the synthesis of halicyclindramide A.

CD spectra of **1a-c** and halicyclindramide A were examined to compare their conformations under different conditions (Figure 2). The overall conformation of **1b** is fairly similar to that of halicyclindramide A in both an aqueous buffer and in trifluoroethanol (TFE). However, for peptide **1a** in an aqueous buffer, stronger intensities were observed at both the negative absorption around 220 nm and the maximum below 200 nm, while only the negative peak was intensified in TFE. The stronger absorptions suggest that the amide group in **1a** forces the molecule to adopt a more stable secondary structure than that produced by the ester and alkane groups in **1b** and **1c**, respectively. This is probably due to the presence of additional H-bonding interaction(s) in **1a**. For the *allo*-Thr analog **1c**, much weaker intensities were observed in both solvents.

Conclusions

In summary, we demonstrated the synthesis of three mimics of halicyclindramide A *via* solid-phase peptide synthesis. The ester linkage between the side-chain of Thr and Sar in halicyclindramide A was replaced by an ester with a different stereochemistry at the Thr residue, an amide, and an alkane. Their solution structures were compared using CD spectroscopy in an aqueous buffer and in TFE. The CD spectra suggested that the amide analog **1a** could adopt a more stable secondary structure, while the structure of the *allo*-Thr analog **1c** was less restricted in both solvents. The conformational stability of **1a** was induced via intramolecular H-bonding interaction(s) of the additional amide linkage and may alter its biological activity greatly by changing the mode of ligand-receptor interaction(s).

Experimental Section

General. All reagents were obtained from commercial suppliers and used without further purification, unless specified. ¹H and ¹³C NMR spectra were collected on a Bruker UI500 spectrometer at resonance frequencies of 500.1 MHz and 125.7 MHz, respectively. The solvents used were DMSO-*d*⁶ and CD₃OD. The chemical shifts are reported in ppm from tetramethylsilane or referenced to the solvent on the δ scale. LRMS (ESI) and HRMS (ES⁺) spectra were recorded on an Agilent 1100 LC/MSD and a Micromass LCT,

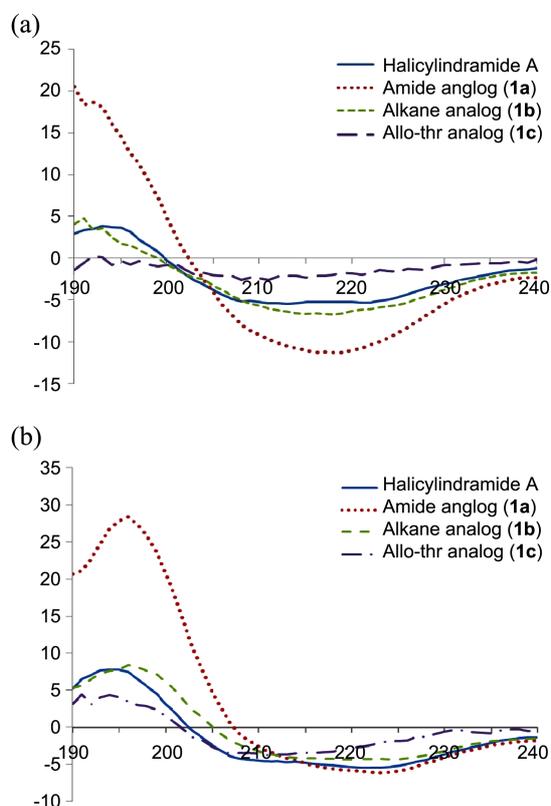


Figure 2. Comparison of CD spectra of **1a-c** and halicyclindramide A in (a) aqueous buffer and (b) TFE.

respectively. Analytical RP-HPLC was carried out using a Kromasil 100-5C18 (4.6 × 150 mm) reverse-phase column at a flow rate of 1.0 mL/min with UV detection at 215 nm and 254 nm. Linear gradients of CH₃CN-H₂O solvents, each containing 0.5% TFA, were used as follows: conditions A (5-100% CH₃CN over 20 min), and conditions B (30-45% CH₃CN over 20 min). For semipreparative HPLC, a Vydac C18 column (5 μm, 10 × 150 mm, 218TP510) was employed at a flow rate of 3.0 mL/min using gradient conditions B.

Peptide Synthesis. Peptide synthesis was accomplished manually *via* stepwise solid-phase synthesis using Fmoc chemistry. Reaction completion was determined using qualitative Kaiser¹³ or chloranil tests¹⁴. Washings between deprotection and coupling were carried out with DMF (7 × 3 min) and CH₂Cl₂ (3 × 3 min) using 10 mL solvent/g resin for each treatment. After removal of the Fmoc group from the Rink amide resin (500 mg, 0.375 mmol, 100–200 mesh, 1% DVB, 0.75 mmol/g), coupling of the amino acids was achieved by shaking the resin in a solution of Fmoc amino acids, 4 equiv of PyBOP (781 mg), 4 equiv of HOBt (230 mg) or HOAt (204 mg), and 6 equiv of DIPEA (0.37 mL) in DMF (3 mL) for 1 h at room temperature. An excess of amino acids [2 equiv for the coupling of Asp, Thr(t-Bu), D-Phe, (*N*-Me)Gln, Dpr(Mtt), Lys(Mtt) and D-Cys(Acm) derivatives and 3 equiv for Arg(Pmc), D-Trp(Boc), Val, D-Val, Pro, (Br)Phe, and D-Ala derivatives] was used. The Fmoc group was removed using 20% (v/v) piperidine in DMF (3 × 5 min).

The Mtt groups in the Dpr and Lys residues were removed by treating the resin with a CH₂Cl₂ solution of TFA and TIS (1% each, 10 × 5 min). Introduction of the Sar residue to the side chain of the Dpr residue was achieved under general conditions using a solution of Alloc-Sar-OH (2.0 equiv), PyBOP (2.5 equiv.), HOBt (2.5 equiv.), and DIPEA (4 equiv.). The Alloc and allyl protecting groups were removed by shaking the resin with 0.1 equiv. of Pd(PPh₄)₄ and 12 equiv. of PhSiH₃ in CH₂Cl₂ (4 mL) under a nitrogen atmosphere for 4 h. The chloranil test was positive, indicating the presence of the secondary amine of the Sar residue.

The cyclization step was carried out using 5 equiv of *N*-methyl morpholine (0.21 mL, 1.88 mmol), 5 equiv of HATU (713 mg, 1.88 mmol), and 1 equiv of HOAt (51 mg, 0.375 mmol) in DMF (3 mL). After shaking overnight, a negative chloranil test result was obtained. A small portion of the resin was cleaved with TFA-CH₂Cl₂ (1:1) and the crude product was analyzed by RP-HPLC. Deprotection of *S*-Acm was performed by treatment with 1 M Hg(OAc)₂ in DMF (pH 4, adjusted with a few drops of acetic acid) for 3 h in darkness. The resin was washed with DMF (3 × 3 min) and β-mercaptoethanol-DMF (1:9, 3 × 3 min) to remove Hg²⁺ from the resin-bound peptide. A small portion of the resin was cleaved as above and the crude product was analyzed by RP-HPLC.

Oxidation of Cys. A performic acid (3%) solution was prepared by mixing 35% hydrogen peroxide and 98% formic acid (8:92, v/v), and the solution was allowed to stand at room temperature for 1 h. The free thiol of the Cys

residue was treated with freshly prepared 3% performic acid (5 mL) in a bath at 0 °C for 2 h under a nitrogen atmosphere. Dimethyl sulfide (0.4 mL) was added to the reaction mixture and the solution was left to stand for 10 min to quench the excess oxidants. A small portion of the resin was cleaved and the crude product was analyzed by RP-HPLC.

Peptide Cleavage and Isolation. The resin-bound peptide was washed thoroughly with DMF (10 × 3 min) and CH₂Cl₂ (5 × 3 min), and then dried *in vacuo* overnight. The peptide was removed from the solid support using TFA containing 2.5% (v/v) *i*-Pr₃SiH and 2.5% (v/v) H₂O for 2 h at room temperature. This cleavage step was repeated three times, and the combined solution was concentrated to ~2 mL under reduced pressure. Cold diethyl ether was added to the solution and the precipitated peptide was filtered and dried to obtain the crude analogs. HPLC and mass spectrometry analyses of the crude product indicated that the major peak was the desired peptide. The crude peptides were dissolved in MeOH and allowed to stand for 2 d at room temperature. After purification by semipreparative RP-HPLC, pure products were obtained as white solids.

The deformyl amide analog: 33 mg, 5% overall yield from the starting resin (500 mg); HRMS (ES⁺) calcd for C₇₆H₁₀₇BrN₂₁Na₂O₂₀S [M + 2Na - H]²⁺ 897.3424, found 897.3483. Analytical RP-HPLC (condition B): *R*_t = 12.7 min.

The deformyl alkane analog: 9 mg, 3% overall yield from the starting resin (230 mg); HRMS (ES⁺) calcd for C₇₆H₁₀₉BrN₂₀NaO₁₉S [M + Na]²⁺ 871.8564, found 871.8555. Analytical RP-HPLC (conditions 30-65% CH₃CN over 20 min): *R*_t = 9.0 min.

The deformyl *allo*-Thr analog: 10 mg, 1.5% overall yield from the starting resin (500 mg); Analytical RP-HPLC (conditions B): *R*_t = 14.6 min.

Formylation. Formylation of the *N*-terminal Ala residue of **1a-c** using cyanomethyl formate was performed using a minor modification of the method of Deutsch and Niclas.¹⁵ Each deformyl analog was dissolved in 1.5 mL of 75% DMF in H₂O containing sodium phosphate buffer (20 mM, pH 7.4). Cyanomethyl formate (10 equiv.) was added to this solution and the mixture was allowed to stand at room temperature for several days in darkness. The reactions were monitored by analytical RP-HPLC and the crude products were purified by semipreparative RP-HPLC to give **1a-c** as white solids. The ¹H NMR and 2D NMR (COSY, ROESY, TOCSY, HMBC, and HSQC) data for **1a-c** were obtained, and their NMR data are listed in the supporting information.

The amide analog (**1a**): 13.5 mg (90% yield based on HPLC analysis); HRMS (ES⁺) calcd for C₇₇H₁₀₈BrN₂₁Na₂O₂₁S [M + Na₂]²⁺ 911.8438, found 911.8446. Analytical RP-HPLC (conditions B): *R*_t = 14.3 min.

The alkane analog (**1b**): 6.8 mg (75% yield); HRMS (ES⁺) calcd for C₇₇H₁₁₀BrN₂₀NaO₂₀S [M + H + Na]²⁺ 886.3578, found 886.3517. Analytical RP-HPLC (conditions A): *R*_t = 13.2 min.

The *allo*-Thr analog (**1c**): 6 mg (60% yield); HRMS (ES⁺) calcd for C₇₈H₁₀₉BrN₂₀Na₂O₂₂S [M + Na₂]²⁺ 919.3437, found 919.3119. Analytical RP-HPLC (conditions B): *R*_t = 15.1 min.

CD Measurements. CD experiments were performed at room temperature using a Jasco 715 spectropolarimeter with a path length of 1.0 cm. Each sample was dissolved in TFE or sodium phosphate buffer (20 mM, pH 7.4) to give a 100 μ M concentration. Data were collected from 260 nm to 190 nm in 1-nm increments at a scan rate of 20 nm/min and a 3-s signal averaging time.

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References

- (a) Li, H.-Y.; Matsunaga, S.; Fusetani, N. *J. Med. Chem.* **1995**, *38*, 338-343. (b) Li, H.-Y.; Matsunaga, S.; Fusetani, N. *J. Nat. Prod.* **1996**, *59*, 163-166.
 - (a) Matsunaga, S.; Fusetani, N.; Konosu, S. *Tetrahedron Lett.* **1984**, *25*, 5165-5168. (b) Matsunaga, S.; Fusetani, N.; Konosu, S. *J. Nat. Prod.* **1985**, *48*, 236-241. (c) Matsunaga, S.; Fusetani, N.; Konosu, S. *Tetrahedron Lett.* **1985**, *26*, 855-856. (d) Ryu, G.; Matsunaga, S.; Fusetani, N. *Tetrahedron Lett.* **1994**, *35*, 8251-8254. (e) Ryu, G.; Matsunaga, S.; Fusetani, N. *Tetrahedron* **1994**, *50*, 13409-13416.
 - Feng, Y.; Carroll, A. R.; Pass, D. M.; Archbold, J. K.; Avery, V. M.; Quinn, R. J. *J. Nat. Prod.* **2008**, *71*, 8-11.
 - Laird, D. W.; LaBarbera, D. V.; Feng, X.; Bugni, T. S.; Harper, M. K.; Ireland, C. M. *J. Nat. Prod.* **2007**, *70*, 741-746.
 - Sato, K.; Horibe, K.; Amano, K.-i.; Mitusi-Saito, M.; Hori, M.; Matsunaga, S.; Fusetani, N.; Ozaki, H.; Karki, H. *Toxicon*. **2001**, *39*, 259-264.
 - Seo, H.; Lim, D. *The Journal of Organic Chemistry* **2009**, *74*, 906-909.
 - Guibé, F. *Tetrahedron* **1998**, *54*, 2967-3042.
 - Vojtkovsky, T. *Pept. Res.* **1995**, *8*, 236-237.
 - Matthiesen, R.; Bauw, G.; Welinder, K. G. *Anal. Chem.* **2004**, *76*, 6848-6852.
 - Greene, T. W.; Wuts, P. G. M. *Protecting Groups in Organic Synthesis*, 3rd ed.; John Wiley & Sons: New York, 1999; Vol. II.
 - Hirs, C. H. W. In *Methods in Enz.*; Hirs, C. H. W., Ed.; Academic Press: New York, 1967; pp 197-199.
 - Deutsch, J.; Niclas, H.-J. *Synthetic Commun.* **1993**, *23*, 1561-1568.
 - Kaiser, E.; Colescott, R. L.; Bossinger, C. D.; Cook, P. I. *Anal. Biochem.* **1970**, *34*, 595-598.
 - Vojtkovsky, T. *Pept. Res.* **1995**, *8*, 236-237.
 - Deutsch, J.; Niclas, H.-J. *Synthetic Commun.* **1993**, *23*, 1561-1568.
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