Stereochemistry Structure of Odonicin

Bao-Lin Li,^{†,‡,*} Jin Li,[†] Ling Tong,[†] Yuan-Jiang Pan,[‡] and Kai-Bei Yu[§]

[†]School of Chemistry and Material Science, Shaanxi Normal University, Xian, Shaanxi 710062, China [‡]Department of Chemistry, Zhejiang University, Hangzhou, Zhejiang 310027, China [§]Chengdu Institute of Organic Chemistry, Academia Sinica, Chengdu, Sichuan 610041, China Received July 5, 2003

Key Words : Odonicin, Ent-kaurene diterpenoid, Isodon henryi, X-ray diffraction analysis

Ent-kaurene diterpenoids have various biological activities, such as antitumor, antibacterial, anti-HIV activities etc.¹⁻³ The plant of the genus *Isodon* used in Chinese traditional medicine⁴ is a major source of compounds of this structural class.⁵ Our investigations on the bitter diterpenoids constituents of the genus *Isodon* led to the first isolation of odonicin (1) from the leaves and tender branches of *Isodon henryi* obtained from Taibai Mountain, Shaanxi Province, P. R. China.⁶



Odonicin, an *ent*-kaurene type diterpenoid, was reported in 1973⁷ as a new compound isolated from *Isodon japonicus*. Up to now, no direct evidence has been reported for the configuration, conformation and crystal structure of compound **1**. Herein, we present the stereochemistry characters of this compound on the basis of X-ray diffraction analysis.

The chemical structure of odonicin is 6β , 15β -diacetoxy-7,20-epoxy-7 β -hydroxy-*ent*-kaur-2,16-dien-1-one. The structure's character is presented clearly in Figure 1 as determined by X-ray diffraction analysis. The molecule of (1) consists of three six-membered rings A, B, C and one five-membered ring D. The bond lengths O2-C1, C1-C2 and C2-C3 are 1.214(3), 1.455(5) and 1.326(4) Å. The bond angles of O2-C1-C10, C2-C1-C10, C1-C2-C3, C2-C3-C4 and C3-C4-C5 are 120.7(3)°, 118.1(2)°, 122.2(3)°, 122.8(3)° and 107.5(2)°, respectively. These values indicate that the C1, C2 and C3 are in sp^2 hybrid, the C1, C2, C3, C4, C10 are in a common plane, and the C5 rises from the common plane. In addition, a conjugate system of α,β -unsaturated ketone exists in ring A, and ring A approximates a boat conformation. Ring B, fused with ring A in anti-form, is in a boat conformation due to the presence of 7,20-epoxy bond. The acetoxyl group at C6 and hydroxyl at C7 are in β - orientations. Ring C fuses with ring B in cis-form and adopts a boat conformation. Ring D is in an envelope conformation, the C14 α -oriented from the common plane of C8, C15, C16 and C13. An acetoxyl group is at C15 in β -orientation. The bond angles of 124.8(3)° (C17-C16-C15), 127.8(3)° (C17-C16-C13), 107.3(2)° (C15-C16-C13) and the bond length of 1.319(5) Å (C16-C17) show the C16 and C17 are sp² hybrid and an exo-methylene attaches at C16. The bond angle of 97.4(2)° (C15-C8-C14) and the bond angles at C16 diverge the normal bond angles of Csp³ and Csp², which leads to the torsion angles at C8 and C16 and a bigger tensile force in ring D. Previous studies on ent-kaurene type diterpenoids indicated their various biological activities^{8,9} can be attributed to the chemical activity of ring D and exo-methylene, which opens ring D and shows the addition reaction of C=C when it interacts with a biological substance.¹⁰ The structure character of (1) suggests its biological activity.

In the crystal packing of compound **1**, molecules arrange as illustrated in Figure 2. The crystal of **1** is a monoclinic crystal system. Space group: P2₁. Accurate lattice constants: a = 9.965(1) Å, b = 11.212(1) Å, c = 10.636(3) Å, $\alpha = 90^{\circ}$, $\beta = 113.50^{\circ}$, $\gamma = 90^{\circ}$. F(000): 460. Cell volume: 1089.8(2) Å³, Z=2. Extinction coefficient: 0.024(3). Two intermolecular hydrogen bonds are present between the hydroxyl H atom at C7 and the carbonyl O atom of the acetoxyl group at C15, the hydroxyl O atom at C7 and the methyl H atom of



Figure 1. Molecular structure of odonicin. H atoms have been omitted for clarity.

^{*}Corresponding author. Phone: +86-29-5303858, Fax: +86-29-5307774, e-mail: baolinli@snnu.edu.cn

Notes



Figure 2. Crystal packing of odonicin.

the acetoxyl group at C15, with $H \cdots O$ separations of 2.861(3) and 3.298(3) Å, respectively. The molecules arrange in a one-dimension structure along the b axis in the crystal resulting from the intermolecular hydrogen bonds.

Experimental Section

General Methods. The ¹³C NMR spectrum of compound **1** was recorded on a Bruker Avance-DMX500 instrument. Chemical shift values δ are given in ppm using TMS as the internal standard. IR spectrum was recorded in KBr pellet on a 170SX FT-IR instrument. X-ray diffraction analysis was performed on a Siemens P4 four-circle diffractometer.

Extraction and isolation. The dried powdered leaves and new branches of *Isodon henryi* (5.0 kg) were extracted with 95% EtOH (15 L × 3) at room temperature for 7 days. After removal of the solvent in vacuo, the residue was suspended in H₂O and extracted with petroleum ether (3 L × 3) and EtOAc (3 L × 3), respectively. The EtOAc extract (122 g) was subjected to column chromatography on silica gel (2 kg, 200-300 mesh) and eluting with gradient mixtures of CHCl₃ and Me₂CO (CHCl₃/Me₂CO: from 10 : 0 to 0 : 10) to afford the crude compound (1). Final purification was accomplished by silica column chromatography to yield (1) (113 mg, 0.0023%). Colorless prism crystals were obtained from a solution of the compound in methanol by slow evaporation at room temperature.

Odonicin (1). $C_{24}H_{30}O_7$, IR v (cm⁻¹): 3447, 2962, 2839, 2867, 1730, 1663, 1377, 1257, 1230, 1143, 1062. ¹³C NMR (CDCl₃) δ (ppm): 197.0 (C-1), 127.9 (C-2), 159.3 (C-3), 35.5 (C-4), 50.8 (C-5), 74.7 (C-6), 96.4 (C-7), 51.3 (C-8), 42.4 (C-9), 46.5 (C-10), 17.5 (C-11), 31.7 (C-12), 35.3 (C-13), 26.0 (C-14), 74.0 (C-15), 157.3 (C-16), 109.4 (C-17), 29.6 (C-18), 24.7 (C-19), 65.3 (C-20), 173.4, 170.2, 21.0,

| Table 1. | Data | collection | and | structure | refinement | for | odonicin | 2 |
|----------|------|------------|-----|-----------|------------|-----|----------|---|
| | | | | | | | | |

| Tuble 1. Data concerton and st | fucture reminiment for subment |
|------------------------------------|--|
| Empirical formula | $C_{24}H_{30}O_7$ |
| Formula weight | 430.48 |
| Temperature | 295(2) K |
| Wavelength | 0.71073 Å |
| Density (calculated) | 1.312 Mg/m ³ |
| Absorption coefficient | 0.096 mm^{-1} |
| Crystal size | $0.54 \times 0.46 \times 0.46 \text{ mm}$ |
| θ range for data collection | 2.09° to 28.49° |
| Limiting indices | $0 \le h \le 13, 0 \le k \le 15, -14 \le l \le 13$ |
| Reflections collected | 3164 |
| Independent reflections | 2898 ($R_{int} = 0.0109$) |
| Absorption correction | None |
| Refinement method | Full-matrix least-squares on F ² |
| Data / restraints / parameters | 2898 / 2 / 289 |
| Goodness-of-fit on F^2 | 0.940 |
| Final R indices $[I > 2\sigma(I)]$ | R1=0.0399, wR2=0.0877 |
| R indices (all data) | R1=0.0589, wR2=0.0945 |
| Absolute structure parameter | 1.2(11) |
| Largest diff. peak and hole | 0.179 and -0.145 e ${\rm \AA}^{-3}$ |

* $(\Delta/\sigma)_{max} = 0.00$, w = $1/[\sigma^2(F_o^2) + (0.0534P)^2]$, where P = $(F_o^2 + 2F_c^2)/3$. 25 Reflection determined crystal cell (2.89 < θ < 14.96), crystal structure was obtained by direct method. The H atom of O5 was synthesized by residual quantity fourier, and data for other H atoms were obtained from theoretical calculations. Siemens SHELXTL 5.10 program was used for all calculation.

| Table 2 | . Selected | geometric | parameters (| (Å, ° |) for odonicir | 1 |
|---------|------------|-----------|--------------|-------|----------------|---|
|---------|------------|-----------|--------------|-------|----------------|---|

| | 8 | | |
|------------|------------|---------------|----------|
| O2-C1 | 1.214(3) | C13-C14 | 1.534(4) |
| C1-C2 | 1.455(5) | C13-H13 | 0.9800 |
| C1-C10 | 1.524(3) | C15-C16 | 1.509(3) |
| C2-C3 | 1.326(4) | C16-C17 | 1.319(5) |
| C2-H2 | 0.9300 | C8-C14 | 1.544(3) |
| C3-C4 | 1.501(4) | C8-C15 | 1.534(4) |
| С3-Н3 | 0.9300 | C17-H17A | 0.9300 |
| C4-C18 | 1.527(5) | C17-H17B | 0.9300 |
| C13-C16 | 1.524(5) | | |
| O2-C1-C2 | 121.2(3) | C7-C8-C14 | 110.9(2) |
| O2-C1-C10 | 120.7(3) | C15-C8-C9 | 110.7(2) |
| C2-C1-C10 | 118.1(2) | C16-C13-C14 | 102.1(2) |
| C3-C2-C1 | 122.2(3) | C13-C14-C8 | 101.1(2) |
| С3-С2-Н2 | 118.9 | C16-C15-C8 | 103.0(2) |
| C1-C2-H2 | 118.9 | C17-C16-C15 | 124.8(3) |
| C2-C3-C4 | 122.8(3) | C17-C16-C13 | 127.8(3) |
| С2-С3-Н3 | 118.6 | C15-C16-C13 | 107.3(2) |
| C4-C3-H3 | 118.6 | C16-C17-H17A | 120.0 |
| C15-C8-C7 | 117.62(18) | C16-C17-H17B | 120.0 |
| C15-C8-C14 | 97.4(2) | H17A-C17-H17B | 120.0 |

21.6 (2 \times AcO-). Data collection and structure determinations of X-ray diffraction analysis are listed in Table 1. Some selected bond lengths and angles are listed in Table 2. Crystallographic data have been deposited with the Cambridge Crystallographic Data Center. CCDC deposition number is 220128.

306 Bull. Korean Chem. Soc. 2004, Vol. 25, No. 2

References

- 1. Nagao, Y.; Ito, N.; Kohno, T.; Kuroda, H.; Fujita, E. Chem. Pharm. Bull. 1982, 30(2), 727-729.
- Wu, Y. C.; Hung, Y. C.; Cheng, F. R.; Cosentino, M.; Wang, H. K.; Lee, K. H. J. Nat. Prod. 1996, 59, 635-637.
- Chen, K.; Shi, Q.; Fujioka, T.; Nakano, T.; Hu, C. Q.; Jin, J. Q.; Kilkuskie, R. E.; Lee, K. H. *Bioorg. Med. Chem.* 1995, *3*, 1345-1348.
- 4. Li, B. L.; Pan, Y. J.; Pan, W. J. Helvetica Chimica Acta 2001, 84(11), 3418-3422.
- 5. Fujita, E.; Node, M. Progress in the Chemistry of Organic Natural Products 1984, 46, 77.
- Li, B. L.; Pan, Y. J.; Yu, K. B. *Tetrahedron Letters* 2002, 43, 3845-3847.
- Fujita, E.; Taoka, M.; Nagao, Y.; Fujita, T. J. Chem. Soc. Perkin Trans. 1 1973, (16), 1760-1765.
- 8. Liu, C. J.; Zhao, Z. H. Chinese Pharm. J. 1998, 33(10), 577-581.
- Watson, W. H.; Zabel, V. Acta Crystallogr., Sect. B 1982, 38(5), 1608-1610.
- Chen, Y. Z.; Wu, Z. W.; Cheng, P. Y. Chinese J. Organic Chemistry 1987, 7(1), 21-28.